

Goa Board of Secondary & Higher Secondary Education
Alto, Betim – Goa
HSSC Practical Examination
CHEMISTRY

Date:
Session: Morning/Afternoon

Time: 3 Hours
Max. Marks: 25

N. B.

- 1) On your answer books write your Examination Seat number and your Laboratory table number.
- 2) Get the burette reading and confirmatory tests initialed by one of the Examiners.
- 3) Check if the number on (i) your table (ii) answer script and (iii) the containers A, B, C, D and F are the same. If not, report immediately to the Examiners.
- 4) Use of non - programmable calculator is allowed.

Atomic Masses:- H=1, C=12, N=14, O=16, S=32, K=39, Mn=55, Fe= 56.

SECTION – I

Q.1. You are provided with two solutions as follows:-

Container A: _____N/M stock solution of hydrated Oxalic acid / Mohr's salt.

Container B: KMnO₄ Solution

Using the stock solution from Container **A**, prepare 100 mL of _____N/M hydrated Oxalic acid /Mohr's salt in the given Standard Measuring flask **C**.

Using the solution prepared in flask **C**, determine N/M of the solution in container **B**.

Also calculate.

- The strength of the solution in container **B** in terms of grams per _____**mL**.
- The percentage purity of the solution in container **B**, _____ **g** of which have been dissolved per _____**mL**. (7 Marks)

Q.2. Determine the functional group of the organic compound supplied to you in Container **F** bearing your table number. Give a complete report of all the tests performed. (2 Marks)

Q.3. Journal + Viva (2+2 Marks)

SECTION – II

Q.4. Analyse the inorganic salt given in container **D** bearing your table number qualitatively and detect the cation and anion present. Give a complete report of all the tests performed. Write the formula of the Compound detected. (8 Marks)

Q.5. Project + Viva (2+2 Marks)